## THE SYNTHESIS OF USEFUL CHIRAL PROSTANOID INTERMEDIATES AND NATURALLY OCCURRING PROSTAGLANDINS FROM AUCUBIN

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The synthesis of optically active intermediate  $\underline{9}$  for prostanoids from aucubin  $\underline{1}$  was described, in which stepwise cleavage of intramolecular cyclic acetal of tetrahydroanhydroaucubigenin  $\underline{2}$  by ethanethiol in the presence of boron trifluoride was successfully accomplished. The synthesis of naturally occurring prostaglandins was also described.

Previously we have reported  $^1$  a stereocontrolled synthesis of optically active prostanoids from aucubin  $\underline{1}^2$ , which includes Lewis acid promoted cross aldol-type condensation between acetal and enol acetate. Herein we wish to describe alternative synthetic approach to prostanoids from aucubin  $\underline{1}$ .

The known tosylate  $\underline{3}^{3,4,5)}$  was treated with excess ethanethiol in dimethoxyethane in the presence of boron trifluoride etherate(2.4 equiv) at 20°C to give the monothioacetal  $\underline{4}^{4)}$  in quantitative yield. Cornforth oxidation of  $\underline{4}$  produced the lactone  $\underline{5}^{4,5)}$  (47% yield, mp 98-100°C,  $[\alpha]_D^{23}$  +218°(c 1.00, chloroform)), which was treated with boron trifluoride etherate(2.0 equiv) in ethanethiol at 20°C to produce the dithioacetal  $\underline{6}^{3)}$  quantitatively. Benzoylation of  $\underline{6}$  afforded the corresponding benzoate  $\underline{7}^{4,5)}$  (mp 79.5-80°C,  $[\alpha]_D^{23}$  -40°(c 1.04, chloroform)) in 75% yield from 5.

Treatment of the benzoate  $\underline{7}$  with N-chlorosuccinimide-silver nitrate in acetonitrile-water<sup>6)</sup> at 0°C gave a mixture of the aldehyde  $\underline{8}$  (more polar) and the epimer  $\underline{9}$  (less polar). The mixture was converted upon treatment with potassium acetate in methanol into the stable isomer  $\underline{9}^{4,5}$  (68% yield from  $\underline{7}$ , mp 132-133.5°C,  $[\alpha]_D^{24}$  -24°(c 1.00, chloroform), NMR:aldehyde proton  $\delta$ =9.78 ppm(d)). This aldehyde

is a useful intermediate for the synthesis of ll-deoxy-ll $\alpha$ -hydroxymethyl prostaglandins. For example, ll-deoxy-ll $\alpha$ -hydroxymethyl PGF $_{2\alpha}^{\phantom{2\alpha}}$  ([ $\alpha$ ] $_{D}^{25}$  +23°(c 0.26, tetrahydrofuran)) was derived from  $\underline{9}$  using essentially the same experimental conditions as those of the conventional methods $^{8}$ ).

Synthesis of naturally occurring prostaglandins was accomplished in the following way. Condensation of the aldehyde 9 with sodio derivatives of dimethyl 2-oxoheptylphosphonate in dimethoxyethane at 20°C produced the trans-enone  $10^{4}$ ) (88% yield,  $[\alpha]_D^{25}$  -25°(c 1.312, chloroform)). Treatment of the enone  $\underline{10}$  with excess zinc borohydride in dimethoxyethane at 20°C afforded a mixture of the (15S) alcohol  $11^{4}$  (less polar) and the (15R) epimer  $12^{4}$  (more polar). These were readily separated by column chromatography on silica gel using ethyl acetate as eluent  $(11; 47\% \text{ yield, } [\alpha]_D^{24} -36\% \text{(c 1.1, chloroform), } 12; 33\% \text{ yield, } [\alpha]_D^{24} -56\% \text{(c 1.3, chloroform)}$ chloroform)). The desired (15S) alcohol  $\underline{11}$  was converted into the tetrahydropyranyl(THP) derivative 134) using dihydropyran(5 equiv) in dichloromethane containing catalytic amount of p-toluenesulfonic acid. Methanolysis of 13 with an equimolar amount of potassium carbonate in methanol at 20°C gave the alcohol  $\underline{14}^{4)}$ , which was converted by the reaction with Cornforth reagent  $^{9)}$  at 20°C to the carboxylic acid  $15^4$ ). The carboxylic acid 15 was converted into  $17^4$ ) via  $16^4$ ) by the known procedure  $^{10}$ ) (condensation with m-chloroperbenzoic acid, followed by decarboxylative rearrangement).

Methanolysis of <u>17</u> with potassium carbonate in methanol at 20°C followed by hydrolysis in acetic acid-water(2:1) at 40°C afforded the diol  $\underline{18}^{4,5}$ ) (20% yield from <u>11</u> after chromatographic purification(silica gel, ethyl acetate)). The diol  $\underline{18}$ , thus obtained, exhibited the same IR and NMR spectra, and thin layer chromatographic behavior in all respects as those of an authentic sample which was prepared according to the conventional method<sup>8)</sup>, and had  $[\alpha]_D^{23}$  -7.1°(c 0.491, chloroform) as compared with  $[\alpha]_D^{27}$  -7.0°(c 1.43, chloroform) found for an authentic sample <sup>8b)</sup>.

The synthesis of naturally occurring forms of prostaglandins from the diol  $\underline{18}$  has been reported by E.J. Corey  $et\ \alpha l.^{8b)}$ , and the synthesis of the diol  $\underline{18}$  constitutes indirect total synthesis of naturally occurring prostaglandins.

$$\frac{6}{2}$$
 R=H
$$\frac{6}{2}$$
 R=C<sub>6</sub>H<sub>5</sub>CO
$$\frac{8}{2}$$
 SC<sub>2</sub>H<sub>5</sub>

 $SC_2H_5$ 

4

R=H

$$\begin{array}{c} H \\ \downarrow 0 \\ H \\ \downarrow 0 \\ H \\ \downarrow 0 \\ SC_2H_5 \end{array}$$

## References and Notes

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